Epoxide Opening-Induced Tandem 8-Azabicyclo[3.2.1]octane to 6-Azabicyclo[3.2.1]octane Rearrangement—Iminium Allylation: Synthesis of (\pm) -Peduncularine

David M. Hodgson, *,† Ruth E. Shelton, † Thomas A. Moss, † and Mouloud Dekhane ‡

Department of Chemistry, University of Oxford, Chemistry Research Laboratory, Mansfield Road, Oxford OX1 3TA, U.K., and AstraZeneca, Silk Road Business Park, Charter Way, Macclesfield SK10 2NA, U.K.

david.hodgson@chem.ox.ac.uk

Received April 25, 2010

LETTERS 2010 Vol. 12, No. 12 2834–2837

ORGANIC



An efficient Lewis acid induced nitrogen-driven rearrangement iminium-trapping cascade from an epoxytropinone 3 gives a 7-allylated 6-azabicyclo[3.2.1]octan-3-one 2, which is converted into the alkaloid (\pm)-peduncularine (1).

(–)-Peduncularine (1) (Scheme 1), the first *Aristotelia* alkaloid isolated, was originally obtained from the Tasmanian shrub *A. peduncularis* and assigned by Bick and co-workers as an indole-pyrrolizidine in 1971;¹ degradative and more detailed spectroscopic studies led them to propose the correct structure in 1979.² Although its biosynthetic precursors are, in common with many indole alkaloids, considered to be tryptamine and a terpenoid unit,³ peduncularine exhibits an unusual 6-azabicyclo[3.2.1]oct-3-ene core with a 7-*exo*-(indol-3-yl)methylene substituent and has proved a popular target for synthetic studies. Klaver, Hiemstra, and Speckamp

reported the first synthesis of (–)-peduncularine in 1989, which unambiguously confirmed the structural reassignment;⁴ however, stereochemistry at C-7 was not controlled. Four formal syntheses have subsequently appeared targeting intermediates in this route⁵ with one, reported by Roberson and Woerpel, being later extended to a racemic total synthesis and which introduced the C-7 stereocenter will full control.⁶ Kitamura and co-workers have recently reported another total synthesis of (±)-peduncularine.⁷ A variety of strategies have been used in the above studies to form the bicyclic core of

[†] University of Oxford.

[‡] AstraZeneca.

⁽¹⁾ Bick, I. R. C.; Bremner, J. B.; Preston, N. W.; Calder, I. C. J. Chem. Soc., Chem. Commun. 1971, 1155–1156.

⁽²⁾ Ros, H.-P.; Kyburz, R.; Preston, N. W.; Gallagher, R. T.; Bick,
I. R. C.; Hesse, M. *Helv. Chim. Acta* 1979, 62, 481–487.
(3) (a) Bick, I. R. C.; Hai, M. A. In *The Alkaloids*; Brossi, A., Ed.;

^{(3) (}a) Bick, I. R. C.; Hai, M. A. In *The Alkaloids*; Brossi, A., Ed.; Academic Press: London, 1985; Vol. 24, pp 113–151. (b) Borschberg, H.-J. In *The Alkaloids*; Cordell, G. A., Ed.; Academic Press: London, 1996; Vol. 48, pp 191–248.

⁽⁴⁾ Klaver, W. J.; Hiemstra, H.; Speckamp, W. N. J. Am. Chem. Soc. 1989, 111, 2588–2595.

^{(5) (}a) Rigby, J. H.; Meyer, J. H. *Synlett* **1999**, *S1*, 860–862. (b) Roberson, C. W.; Woerpel, K. A. *Org. Lett.* **2000**, *2*, 621–623. (c) Lin, X.; Stien, D.; Weinreb, S. M. *Tetrahedron Lett.* **2000**, *41*, 2333–2337. (d) Washburn, D. G.; Heidebrecht, R. W., Jr.; Martin, S. F. *Org. Lett.* **2003**, *5*, 3523–3525.

⁽⁶⁾ Roberson, C. W.; Woerpel, K. A. J. Am. Chem. Soc. 2002, 124, 11342–11348.

⁽⁷⁾ Kitamura, M.; Ihara, Y.; Uera, K.; Narasaka, K. Bull. Chem. Soc. Jpn. 2006, 79, 1552–1560.





peduncularine:⁸ four approaches used ring closure onto an existing ring (Speckamp, Roberson and Woerpel, Weinreb, and Martin) (Scheme 1), whereas Rigby and Meyer utilized a [6 + 2] cycloaddition,^{5a} and Kitamura formed and then fragmented a tricyclic intermediate.⁷ In the present paper, we communicate a cascade rearrangement process to access a 7-allylated 6-azabicyclo[3.2.1]octan-3-one, which is subsequently converted to peduncularine.

With the electrophile-induced rearrangement of 8-azabicyclo[3.2.1]octa-2,6-dienes to 6-azabicyclo[3.2.1]oct-2-enes reported by Davies and Cao in mind,⁹ we envisaged that readily accessible epoxytropinone 3^{10} (Scheme 2) could un-

Scheme 2. Strategy to Peduncularine



dergo a Lewis acid assisted, nitrogen-driven (formal) 1,2shift (shown in 4) to generate intermediate iminium 5. By analogy to the trapping of a related iminium in Roberson and Woerpel's total synthesis,⁶ intermediate iminium **5** should be capable of interception by allyltrimethylsilane¹¹ to generate the 7-allylated 6-azabicyclo[3.2.1]octan-3-one **2**, possessing potentially suitable functionality for conversion to peduncularine.

So as to examine the above strategy, epoxytropinone **3** was prepared following the procedure of Mann and de Almeida Barbosa¹⁰ with slight modifications.¹² This chemistry involves [4 + 3] cycloaddition between commercially available *N*-methoxycarbonyl pyrrole¹³ and 1,1,3,3-tetrabromoacetone¹⁴ to provide 8-azabicyclo[3.2.1]oct-6-en-3-one **6** (36%), which was subsequently epoxidized with methyl(trifluoromethyl)dioxirane to give *exo*-epoxytropinone **3** (87%, Scheme 3). Pleasingly, we found that epoxytropinone **3**



cleanly underwent the desired rearrangement-allylation to give 6-azabicyclo[3.2.1]octan-3-one **2** (80%) following reaction with TMSOTf (3 equiv), allyltrimethylsilane (1.1 equiv), and 2,6-lutidine (1.1 equiv) in CH₂Cl₂ at rt for 16 h (Scheme 3). Attempts to use cat. TMSOTf resulted in no reaction, whereas other variations to the above conditions (Lewis acid, solvent, temperature, reaction time) were detrimental to the yield. The *exo*-disposition of the allyl group was established by X-ray crystallographic analysis of a diol derived from reduction of ketone **2**.¹² Despite the presence of the *syn* oxy substitutent on the one-carbon bridge of the putative iminium **7**,¹⁵ *exo*-face allylation appeared to be the sole reaction

- A. B.; Kee, A.; Ladduwahetty, T.; Smith, D. F. Chem. Commun. 1990, 1412–1414. (b) Magnus, P.; Ródriguez-López, J.; Mulholland, K.; Matthews,
- I. Tetrahedron 1993, 49, 8059–8072. (c) Yeung, Y.-Y.; Hong, S.; Corey,

⁽⁸⁾ For other strategies to 6-azabicyclo[3.2.1]oct-3-enes, see: (a) Holmes,

E. J. J. Am. Chem. Soc. 2006, 128, 6310-6311.

⁽⁹⁾ Davies, H. M. L.; Cao, G. Tetrahedron Lett. 1998, 39, 5943-5946.

pathway; even on scale-up (using 8 g of epoxytropinone 3) none of the *endo*-addition product was observed.

With the viability of the rearrangement-allylation of epoxytropinone **3** established, attention turned to the introduction of the endocyclic (C-3–C-4) double bond present in peduncularine. We considered that regioselective alkene installation in the desired sense should be possible using an enolization strategy with ketone **2**, due to a potential activating (acidifying) effect of the carbamate functionality. While several avenues based on this theme were investigated, the following sequence was developed to access endocyclic alkene **12** (Scheme 4). Bromination (NBS/NaHCO₃) of a



Scheme 4. Installation of the Endocyclic Alkene from Ketone 2

regioisomeric mixture of crude silyl enol ethers, formed from TBS-protected ketone **8** using LTMP/Me₃SiCl, gave a chromatographically separable mixture of bromoketones **9** (25%) and **10** (51%). The undesired 2-bromoketone **9** could be easily debrominated with Zn/Cu couple to retrieve starting ketone **2** (95%) for reuse. Reduction of the 4-bromoketone **10** using NaBH₄ in MeOH/THF (1:1) at -78 °C followed by careful quenching with 1 M aq HCl at 0 °C gave a single bromohydrin **11** (93%, assigned as the di-*exo*-isomer on the basis of NOE studies¹² and a related reduction in our

synthesis of ibogamine¹⁶). Bromohydrin **11** underwent a Boord-like elimination¹⁷ using Zn dust in AcOH at reflux to give the desired endocyclic alkene **12** (80%); rupture of the bicyclic system arising from elimination toward the carbamate was not observed.

Indole construction from alkene **12** (Scheme 5) employed several transformations similar to those found in





the later stages of Roberson and Woerpel's peduncularine synthesis; their route proceeded through an intermediate differing from alkene 12 at the one-carbon bridge (by virtue of a directly attached (and epimeric) dimethylphenylsilyl group) and with trimethylsilylethoxycarbonyl protection at nitrogen.⁶ Hydroboration/oxidation of the terminal alkene present in alkene 12 gave alcohol 13 (95%), and the derived aldehyde 14, formed (84%) using cat. TPAP/NMO, was converted into the corresponding 1.3-dioxane 15 (70%), the latter for protection during the next two steps and to potentially improve reaction efficiency in the following Fischer indole reaction.¹⁸ Carbamate deprotection with concomitant desilylation of 1,3-dioxane 15 was achieved using KOH in refluxing *i*-PrOH to provide a crude amino alcohol,¹⁹ which underwent reductive amination²⁰ to install the N-*i*-Pr group giving alcohol 16 (78% yield from 1,3-dioxane 15).

⁽¹⁰⁾ Mann, J.; de Almeida Barbosa, L.-C. J. Chem. Soc., Perkin Trans. 1 1992, 787–790.

⁽¹¹⁾ For recent reviews on *N*-acyl iminium chemistry, see: (a) Yazici, A.; Pyne, S. G. *Synthesis* **2009**, 339–368. (b) Yazici, A.; Pyne, S. G. *Synthesis* **2009**, 513–541.

⁽¹²⁾ See Supporting Information for details.

⁽¹³⁾ Hodge, P.; Rickards, R. W. J. Chem. Soc. 1963, 2543-2545.

⁽¹⁴⁾ Kim, H.; Hoffman, H. M. R. *Eur. J. Org. Chem.* 2000, 2195–2220.
(15) Silylenol ether formation from the ketone functionality in epoxytropinone 3 would be reasonably anticipated under the reaction conditions used, and is likely-though not certain (see: Murata, S.; Suzuki, M.; Noyori, R. *Bull. Chem. Soc. Jpn.* 1982, 55, 247–254)-to precede the rearrangement.

⁽¹⁶⁾ Hodgson, D. M.; Galano, J.-M. Org. Lett. 2005, 7, 2221–2224.

^{(17) (}a) Swallen, L. C.; Boord, C. E. J. Am. Chem. Soc. **1930**, 52, 651–660. (b) For a review, see: Schlosser, M. In *Houben-Weyl: Methoden der* organischen Chemie; Thieme Verlag: Stuttgart, 1972; Vol. V/lb, pp 204–219.

⁽¹⁸⁾ Bidylo, T. I.; Yurovskaya, M. A. Chem. Heterocycl. Compd. 2008, 44, 379–418.

Fischer indole synthesis^{18,21} from alcohol **16** furnished indole **17** in 72% yield (Scheme 5). Indole **17** (epimeric at the hydroxyl to an intermediate in Roberson and Woerpel's synthesis)⁶ was oxidized under modified Parikh–Doering conditions,²² to give ketone **18** in 75% yield, completing a formal synthesis. The final methylenation step was repeated using freshly prepared Tebbe reagent,²³ which provided (\pm)peduncularine in 59% yield (lit.⁶ 56%) and with data fully consistent with that previously reported.¹²

(21) Chen, C.-Y.; Senanayake, C. H.; Bill, T. J.; Larsen, R. D.; Verhoeven, T. R.; Reider, P. J. J. Org. Chem. **1994**, 59, 3738–3741.

(22) (a) Parikh, J. R.; Doering, W.; von, E. J. Am. Chem. Soc. 1967, 89, 5505–5507. (b) Chen, L.; Lee, S.; Renner, M.; Tian, Q.; Nayyar, N. Org. Process Res. Dev. 2006, 10, 163–164. (c) Gerasyuto, A. I.; Hsung, R. P. J. Org. Chem. 2007, 72, 2476–2484.

(23) (a) Cannizzo, L. F.; Grubbs, R. H. J. Org. Chem. **1985**, 50, 2386–2381. (b) Roberson, C. W. Ph.D. Dissertation, University of California, Irvine, 2001; pp 129–130.

In summary, a synthesis of (\pm) -peduncularine has been completed in 15 steps from commercially available materials. The key transformation is an efficient Lewis acid induced nitrogen-driven rearrangement iminium-trapping cascade from an easily accessible epoxytropinone **3** to give the more unusual 6-azabicyclo[3.2.1]octane system present in the alkaloid **1**. Studies to render the approach asymmetric, by enantioselective desymmetrization of the achiral epoxytropinone **3**, are in progress and will be reported in due course.

Acknowledgment. We thank the EPSRC and AstraZeneca for a CASE award (to R.E.S), and Dr. A. Cowley (Oxford) for performing X-ray analysis. We also thank Prof. K. A. Woerpel (Irvine) for his correspondence concerning methylenation of ketone 18.

Supporting Information Available: Experimental procedures for the preparation of new compounds and characterization data, including ¹H and ¹³C NMR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

OL100943J

⁽¹⁹⁾ Extraction using 10% *i*-PrOH in CHCl₃ (*The Synthetic Organic Chemist's Companion*; Pirrung, M. C., Ed.; John Wiley & Sons: New York, 2007; p 109) was found to be essential to ensure a good recovery of the crude polar amino alcohol.

⁽²⁰⁾ Abdel-Magid, A. F.; Carson, K. G.; Harris, B. D.; Maryanoff, C. A.; Shah, R. D. *J. Org. Chem.* **1996**, *61*, 3849–3862.